

## Chemistry Chapter 10 Study Guide

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dozen 4. gross 5. 200 6. ream 7. 6,000,000,000 8. 0.5 mol 9. 6.02 10<sup>23</sup> 10. four moles 11. 6.02 10<sup>23</sup> 1 mol Cu 12. 4 23 4 1 mol CH 6.02 10 molecules CH 13. 23 1 mol Xe 6.02 10 molecules Xe 14. 23 2 ...

### Ch 10 Study Guide TE

Chapter 10 Study Guide. How many moles are in 34.6 L of BaO? How many grams are in 4.6 moles of PbSO<sub>4</sub>? How many molecules are in 6.5 x 10<sup>3</sup> moles of CO<sub>2</sub>? If you have 2.68 moles of S<sub>2</sub>O<sub>3</sub>, how many atoms of S do you have? How many liters are in 4.39 x 10<sup>22</sup> atoms of Mg? The compound methyl butanoate smells like apples.

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